### **Version Marked To Show Changes**

#### IN THE SPECIFICATION

### Please replace the paragraph beginning at 1:4 with the following rewritten paragraph:

This application [claims benefit under 35 U.S.C. §119 of provisional application USSN 60/113,130, filed on December 21, 1998, which is] is a divisional of Ser. No. 09/467,536, filed December 20, 1999, pending, which claims the benefit of U.S. Provisional Patent Application No. 60/113,130, filed December 21, 1998, abandoned, the entire disclosures of which are herein incorporated by reference in [its] their entirety for all purposes.

### Please replace the paragraph beginning at 11:17 has been amended as follows:

Figures 4A, 4B, 4C, and 4D illustrate altered specificity patterns relative to WT as the level of negative charge increases in N62C, L271C, S156C and S166C mutants and CMMs with suc-AAPF-pNA as the substrate: Figure 4A: The  $k_{cat}/K_M$ s for N62C CMMs alternate at moderately reduced levels, 1.5- to 3.5-fold lower than WT, which are established by the initial mutation to N62C (R=H). Figure 4B: L217C CMMs show steady but lower levels of  $k_{cat}/K_M$ , 4-to 5.5-fold lower than WT, which are again established by the initial mutation to cysteine. The exception is L217C-c which is only 2.5-fold lower than WT, possibly due to favorable binding of substrate to the phenyl ring of the aromatic side chain introduced by modification. Figure 4C: From the small reduction caused by mutation to S156C (R=H),  $k_{cat}/K_M$ s decrease monotonically to 6-fold lower than WT for S156C-d. The  $k_{cat}/K_M$  of S156C-e is partially restored. Figure 4D:  $k_{cat}/K_M$  decreases only 2.5-fold upon mutation to S166C (R=H) but decreases dramatically to 11-fold lower than WT when the negatively charged sulfonatoethyl side chain a is introduced. In parallel to N62C and L217C CMMs,  $k_{cat}/K_M$  for S166C CMMs does not decrease further to any significant extent as the level of negative charge increases. R group key: a = sulfonatoethyl; b = 4-carboxybutyl; c = 3,5-dicarboxybenzyl; d = 3,3-dicarboxybutyl; e = 3,3,4-tricarboxybutyl.

# Please replace the paragraph beginning at 12:17 has been amended as follows:

Figures 8A, 8B, 8C and 8D show altered specificity patterns for N62C (Fig. 8A), L271C (Fig. 8B), S156C (Fig. 8C), and S166C (Fig. 8D) CMMs: variations in  $\ln(k_{cat}/K_M)$ , with suc-AAPF-pNA as the substrate, for cysteine mutants and positively charged CMMs relative to WT. R group key: a = 2-aminoethyl; b = 2-(trimethylammonium)ethyl; c = 5,5-bis(aminomethyl)-3-oxo-hexyl; d = 2,2-bis(aminomethyl)-3-aminopropyl.

#### Please replace the paragraph beginning at 14:25 has been amended as follows:

Preferred serine hydrolases comprising this invention include the trypsin-chymotrypsin proteases, the subtilisin proteases, and the alpha/beta hydrolases. In a particularly preferred embodiment the enzyme is protease, more preferably a subtilisin (e.g. a Bacillus [lentis] lentus subtilisin). Subtilisin is a serine endoprotease (MW ~27,500) which is secreted in large amounts from a wide variety of Bacillus species. The protein sequence of subtilisin has been determined

from at least four different species of Bacillus (see, e.g., Markland et al. (1971) pages 561-608 In: The Enzymes, ed. Boyer P. D., Acad Press, New York, Vol. III, pp.; Nedkov et al. (1983) Hoppe-Seyler's Z. Physiol. Chem. 364: 1537-1540). The three-dimensional crystallographic structure of subtilisin BPN' (from B. [amyloligoefaciens] amyloliquefaciens) to 2.5 Å resolution has also been reported (Wright et al. (1969) Nature 221, 235-242; Drenth et al. (1972) Eur. J. Biochem. 26: 177-181. These studies indicate that although subtilisin is genetically unrelated to the mammalian serine proteases, it has a similar active site structure. The x-ray crystal structures of subtilisin containing covalently bound peptide inhibitors (Robertus, et al. (1972) Biochemistry 11: 2439-2449), product complexes (Robertus et al. (1972) Biochemistry 11: 4293-4303), and transition state analogs (Matthews et al (1975) J. Biol. Chem. 250: 7120-7126; Poulos et al. (1976) J. Biol. Chem. 251, 1097-1103), which have been reported have also provided information regarding the active site and putative substrate binding cleft of subtilisin. In addition, a large number of kinetic and chemical modification studies have been reported for subtilisin (Philipp et al. (1983) Mol. Cell. Biochem. 51:5-32; Svendsen (1976) Carlsbera Res. Comm. 41: 237-291; Markland, Id.) as well as at least one report wherein the side chain of methionine at residue 222 of subtilisin was converted by hydrogen peroxide to methioninesulfoxide (Stauffer et al. (1965) J. Biol. Chem. 244: 5333-5338).

### Please replace the Table beginning at 33:21 has been amended as follows:

Table 1. Kinetic Parameters<sup>a</sup> for Modified Enzymes

Entry	Enzyme	Pocket	Rº	Level of Charge	$k_{cat}(s^{-1})$	$K_M$ (mM)	$\frac{k_{cat}/K_M}{(\mathrm{s}^{-1}\mathrm{mM}^{-1})}$
1	SBL-WT	-	-	_	153 ± 4	$0.73 \pm 0.05$	209 ± 15
2	N62C	S <sub>2</sub>	Н	0	174 ± 9	$1.90 \pm 0.20$	92 ± 11
3			a	1	$119 \pm 4$	$0.93 \pm 0.07$	$128 \pm 11$
4			b	1	$106 \pm 2$	$1.01 \pm 0.05$	$105 \pm 6$
5			c	2	$113 \pm 7$	$1.00 \pm 0.10$	$113 \pm 13$
6			d	2	$90 \pm 4$	$1.47 \pm 0.14$	$61 \pm 6$
7			e	3	$129 \pm 3$	$1.46 \pm 0.06$	$88 \pm 4$
8	L217C	S <sub>1</sub> '	Н	0	41 ± 1	$0.80 \pm 0.04$	51 ± 3
9			a	1	$89 \pm 6$	$1.80 \pm 0.20$	$50 \pm 6$
10			b	1	$54 \pm 1$	$1.03 \pm 0.06$	$52 \pm 3$
11			c	2	$69 \pm 2$	$0.81 \pm 0.06$	$85 \pm 7$
12			d	2	$63 \pm 2$	$1.65 \pm 0.11$	$38 \pm 3$
13			e	3	$55 \pm 2$	$1.48 \pm 0.08$	$37 \pm 3$
14	S156C	$S_1$	Н	0	$125 \pm 4$	$0.85 \pm 0.06$	147 ± 11
15			a	1	$87 \pm 2$	$1.20 \pm 0.07$	$73 \pm 5$
16			b	1	$76 \pm 1$	$1.08 \pm 0.04$	$70 \pm 3$
17			c	2	$61 \pm 1$	$1.39 \pm 0.10$	$44 \pm 3$

18		d	2	$53 \pm 1$	$1.67 \pm 0.06$	$32 \pm 1$
19		e	3	$74 \pm 2$	$1.87 \pm 0.08$	$39 \pm 2$
20	S166C	H	0	$42 \pm 1$	$0.50 \pm 0.05$	$84 \pm 9$
21		a	1	$25 \pm 1$	$1.34 \pm 0.08$	$19 \pm 1$
22		Ъ	1	$48 \pm 1^{b}$	$1.52 \pm 0.06$	$31 \pm 1^{b}$
23		c .	2	$47 \pm 3$	$1.60 \pm 0.20$	$29 \pm 4$
24		d	2	$67 \pm 2$	$2.26 \pm 0.10$	$30\pm2$
25		е	3	$76 \pm 2$	$2.46 \pm 0.11$	$31 \pm 2$

<sup>&</sup>lt;sup>a</sup> Michaelis-Menten constants were measured at 25°C according to the initial rates method in 0.1 M Tris-HCl buffer at pH 8.6, 0.005% Tween 80, 1% DMSO, Suc-AAPF-pNA as the substrate.

## Please replace the Table beginning at 48:25 has been amended as follows:

Table 2. Kinetic Parameters<sup>a</sup> for Modified Enzymes

Entry	Enzyme	Pocket	R <sup>b</sup>	Level of Charge	$k_{cat}(s^{-1})$	$K_M$ (mM)	$\frac{k_{cat}/K_M}{(\mathrm{s}^{-1}\mathrm{mM}^{-1})}$
1	SBL-WT	-	-	_	153 ± 4	$0.73 \pm 0.05$	209 ± 15
2	N62C	S <sub>2</sub>	Н	0	174 ± 9	1.90 ± 0.20	92 ± 11
3			a	1	$103 \pm 5$	$1.00 \pm 0.10$	$103 \pm 11$
4			b	1	$73 \pm 2$	$0.86 \pm 0.05$	$85 \pm 5$
5			С	2	$92 \pm 3$	$1.06 \pm 0.07$	$87 \pm 6$
6			d	3	$98 \pm 3$	$1.17 \pm 0.08$	$84 \pm 6$
7	L217C	S <sub>1</sub> '	Н	0	41 ± 1	$0.80 \pm 0.04$	51 ± 3
8			a	1	$38 \pm 1$	$0.64 \pm 0.06$	$59 \pm 6$
9			b	1	$43 \pm 1$	$0.69 \pm 0.03$	$62 \pm 3$
10			С	2	$8.0\pm0.2$	$2.94 \pm 0.28$	$2.7 \pm 0.3$
11			d	3	$23 \pm 3$	$2.90 \pm 0.16$	$7.8 \pm 1.2$
12	S156C	$S_1$	Н	0	125 ± 4	$0.85 \pm 0.06$	$147 \pm 11$
13			a	1	$90 \pm 2$	$0.73 \pm 0.04$	$123 \pm 7$
14			b	1	$68 \pm 2$	$0.74 \pm 0.04$	$92 \pm 5$
15			c	2	$64 \pm 1$	$0.76 \pm 0.04$	$85 \pm 5$

<sup>&</sup>lt;sup>b</sup> Based on total protein concentration. Certain inconsistencies between active enzyme concentration as determined by active site titration with PMSF (Hsia *et al.* (1996) *J. Anal. Biochem.* 242: 221-227) and total protein concentration have been reported for negatively charged mutants of SBL. These are characterized by sluggish fluoride ion concentration bursts and high rates of background PMSF hydrolysis. Active enzyme concentration values for S166C-b were low and gave rise to an anomalous value for kcat (270  $\pm$  5 s-1). Consequently the value shown was calculated using total protein concentration as determined by absorbance at 280 nm (e280 = 23000 M-1 cm-1) (Grøn et al. (1990) Eur. J. Biochem. 194:897-901). The KM value determined (1.52  $\pm$  0.06 mM) is not concentration dependent.

c a = sulfonatoethyl; b = 4-carboxybutyl; c = 3,5-dicarboxybenzyl; d = 3,3-dicarboxybutyl; e = 3,3,4-tricarboxybutyl.

16		d	3	$46 \pm 1$	$0.81 \pm 0.05$	$57 \pm 4$
17	S166C	Н	0	$42 \pm 1$	$0.50 \pm 0.05$	$84 \pm 9$
18		a	1	$50 \pm 1$	$0.68 \pm 0.04$	$74 \pm 5$
19		b	1	$33 \pm 2$	$1.42 \pm 0.13$	$23 \pm 2$
20		c	2	$55 \pm 2$	$1.27 \pm 0.10$	$43 \pm 4$
21		d	3	$9.3 \pm 0.2$	$1.16 \pm 0.05$	$8.0 \pm 0.4$

<sup>&</sup>lt;sup>a</sup> Michaelis-Menten constants were measured at 25° C according to the initial rates method in 0.1 M Tris-HCl buffer at pH 8.6, 0.005% Tween 80, 1% DMSO, Suc-AAPF-pNA as the substrate.

<sup>b</sup> a = 2-aminoethyl; b = 2-(trimethylammonium)ethyl; c = 5,5-bis(aminomethyl)-3-oxo-hexyl; d = 2,2-

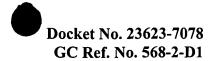
# Please replace the Table beginning at 63:22 has been amended as follows:

Table [1]3: Kinetic Evaluation(a) of Altered S<sub>1</sub> Pocket Specificity

Entry	Enzyme	Substrate	K <sub>M</sub> mM	$k_{\rm cat}  {\rm s}^{-1}$	$k_{cat}/K_{\rm M}$ $s^{-1}$ mM <sup>-1</sup>
1	WT	suc-AAPF-pNA	$0.73 \pm 0.08$	$153 \pm 4$	209 ± 15
2	S166C-S-a	suc-AAPF-pNA	$0.68 \pm 0.04$	$50 \pm 1$	$74 \pm 5$
3	S166C-S-b	suc-AAPF-pNA	$1.34 \pm 0.08$	$25.0 \pm 0.7$	$19 \pm 1$
4	S166C-S-c	suc-AAPF-pNA	$1.17 \pm 0.06$	$23.1 \pm 0.5$	$20 \pm 1$
5	S166C-S-d	suc-AAPF-pNA	$1.6 \pm 0.2$	$47 \pm 3$	$29 \pm 4$
6	S166C-S-e	suc-AAPF-pNA	$1.09 \pm 0.07$	$82 \pm 2$	$75 \pm 5$
7	S166C-S- <b>f</b>	suc-AAPF-pNA	$0.70 \pm 0.05$	$4.8 \pm 0.1$	$6.90 \pm 0.05$
8	S166C-S-g	suc-AAPF-pNA	$0.74 \pm 0.07$	$29 \pm 1$	$41 \pm 4$
9	S166C-S- <b>h</b> (b)	suc-AAPF-pNA	$1.52 \pm 0.06$	$48 \pm 1$	$31 \pm 1$
10	S166C-S- <b>i</b> (b)	suc-AAPF-pNA	$2.26 \pm 0.10$	$67 \pm 2$	$30 \pm 2$
11	S166C-S-j <sup>(b)</sup>	suc-AAPF-pNA	$2.46 \pm 0.11$	$76 \pm 2$	$31 \pm 2$
12	WT	suc-AAPA-pNA	$2.0 \pm 0.1$	$17.7 \pm 0.3$	$8.8 \pm 0.4$
13	S166C-S-c	suc-AAPA-pNA	$0.8 \pm 0.1$	$6.8 \pm 0.3$	$9 \pm 1$
14	S166C-S-e	suc-AAPA-pNA	$1.90 \pm 0.03$	$6.8 \pm 0.4$	$3.6 \pm 0.6$
15	S166C-S- <b>f</b>	suc-AAPA-pNA	$1.90 \pm 0.07$	$28.2 \pm 0.4$	$14.8 \pm 0.6$
16	S166C-S-g	suc-AAPA-pNA	$1.74 \pm 0.04$	$9.65 \pm 0.07$	$5.54 \pm 0.3$
17	WT	suc-AAPR-pNA	$7.2 \pm 0.7$	$0.16 \pm 0.01$	$0.022 \pm 0.002$
18	S166C-S- <b>b</b>	suc-AAPR-pNA	$3.4 \pm 0.3$	$0.17 \pm 0.01$	$0.050 \pm 0.005$
19	S166C-S-d	suc-AAPR-pNA	$5.5 \pm 1.1$	$0.68 \pm 0.08$	$0.12 \pm 0.03$
20	S166C-S-h	suc-AAPR-pNA	$8.2 \pm 0.9$	$0.35 \pm 0.02$	$0.041 \pm 0.005$
21	S166C-S-i	suc-AAPR-pNA	$5.3 \pm 0.5$	$0.43 \pm 0.02$	$0.080 \pm 0.008$
22	S166C-S-j	suc-AAPR-pNA	$5.2 \pm 0.6$	$1.06 \pm 0.07$	$0.20 \pm 0.03$
23	WT	suc-AAPE-pNA	$4.4 \pm 0.4$	$1.75 \pm 0.08$	$0.40 \pm 0.04$
24	S166C-S-a	suc-AAPE-pNA	$1.9 \pm 0.1$	$14.5 \pm 0.3$	$7.6 \pm 0.4$

(a) Michaelis-Menten constants were measured by the initial rates method in pH 8.6 Tris-HCl buffer at 25° C with suc-AAPF-pNA as the substrate.

bis(aminomethyl)-3-aminopropyl.



(b) Based on total protein concentration. Certain inconsistencies between active enzyme concentration as determined by active site titration with PMSF (Hsia et al. (1996) J. Anal. Biochem. 242: 221-227) and total protein concentration have been reported for negatively charged mutants of SBL. These are characterized by sluggish fluoride ion concentration bursts and high rates of background PMSF hydrolysis. Active enzyme concentration values for S166C-S-h, S166C-S-i, and S166C-S-j were low and gave rise to anomalous values for kcat. Consequently the values shown were calculated using total protein concentration as determined by absorbance at 280 nm ( $\varepsilon_{280} = 23000 \text{ M-1 cm-1}$ ) (Grøn et al. (1990) Eur. J. Biochem. 194:897-901). The  $K_M$  values determined are not concentration dependent. (c)  $\mathbf{a} = 2$ -aminoethyl;  $\mathbf{b} = \text{sulfonatoethyl}$ ;  $\mathbf{c} = \text{CH}_2\text{C}_6\text{H}_5$ ;  $\mathbf{d} = \text{CH}_2\text{C}_6\text{H}_4$ -3,5-(COO')2;  $\mathbf{e} = \text{CH}_2(\text{CH}_2)_8\text{CH}_3$ ;  $\mathbf{f} = \text{CH}_2\text{C}_6\text{H}_{11}$ ;  $\mathbf{g} = \text{steroidyl}$ ;  $\mathbf{h} = \text{CH}_2(\text{CH}_2)_2\text{CH}_2\text{COO'}$ ;  $\mathbf{i} = \text{CH}_2\text{C}(\text{CH}_3)(\text{COO'})_2$ ;  $\mathbf{j} = \text{CH}_2\text{C}_4\text{C}(\text{COO'})_3$ .